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The two main types of intermolecular forces that exist between molecules are dipole-dipole forces (including hydrogen bonds) and London dispersion forces.

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DEPARTMENT OF CHEMISTRY & BIOCHEMISTRY. Computational Quantum Chemistry, Wiley , New York (1988). Intermolecular forces (IMFs)

Concept b - intermolecular interactions

The halogen elements create diatomic, non-polar particles. Of the halogens, only iodine is a solid. What type of intermolecular forces are seen for this species?

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