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as implemented in the CADPAC 6.5 program package was used to calculate intermolecular interaction Biopolymers 61: 99 110. Web of A Suite of Quantum

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Possible intermolecular interactions and a interaction used ab initio quantum chemistry methods. The calculations Biochemistry 44:

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Introduction. The interactions between nonpolar molecules are limited to the weakest of intermolecular forces, London Dispersion Interactions, due to the symmetry of

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Buckingham, A. D. (1978) Basic theory of intermolecular forces: Applications to small molecules, in Intermolecular Interactions: From Diatomics to Biopolymers, ed. B

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Various characteristics of mesomorphism can be explained using intermolecular interactions between a pair of liquid crystalline molecules. The intermolecular

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DEPARTMENT OF CHEMISTRY & BIOCHEMISTRY. Computational Quantum Chemistry, Wiley , New York (1988). Intermolecular forces (IMFs)

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which is formulated and calibrated on the basis of quantum chemistry polarizable ligands. for the intermolecular interactions with errors

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Abstract. Long range interactions between the ground state alkali diatomics , , and NaK-NaK are examined. Interaction energies are first determined from ab initio

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of the complexes formed between methyl derivatives of ethylene in Perspectives in Quantum Chemistry and Interactions from Diatomics to Biopolymers,

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The halogen elements create diatomic, non-polar particles. Of the halogens, only iodine is a solid. What type of intermolecular forces are seen for this species?

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